

Structural Preferences of Metal Chalcogenide based Nanowires (MX; M = Au, Ag; X = S, Se): A Computational Study

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This study investigates the structural stability and electronic properties of α -MX nanowires and β -MX nanosheets (M=Au, Ag; X=S, Se) using density functional theory. Our findings confirm the thermal and dynamic stability of all nanostructures, with β -MX phases being more stable than α -MX, albeit with minimal energy differences between the two phases. Electron localization analysis reveals predominantly ionic bonding character between M (Ag and Au) and X (S and Se) atoms. All the phase exhibits metalophilic interactions (Argentophilic and Auophilic), with α -MX nanowires demonstrating a higher level of interaction compared to β -MX nanosheets, attributed to

the curvature inherent in the nanowires. Significantly, these interactions are effectively small and do not compromise the relative stability between the α -MX and β -MX phases. Additionally, both α -MX and β -MX structures exhibit characteristics of indirect bandgap semiconductor behaviour. The band gap values do not significantly differ between α -MX and β -MX phases. Fermi analysis reveals that α -MX phases exhibit small effective electron masses and large hole mobility along the uniaxial direction. These findings suggest promising prospects for α -MX nanowires as 1D semiconductors in nano-electronics and nano-optoelectronics applications in the future.

Introduction

Low-dimensional (D) nanomaterials have drawn much attention among materials scientists because of unique characteristics that cannot be found in bulk or macro-scale materials.^[1] The distinctive properties of low-dimensional nanostructures are inherently linked to their small size and low dimensionality, enabling their utilization in a range of fields, such as photodetectors,^[2] gas sensors,^[3] solar cells,^[4] secondary batteries,^[5] and electrocatalysts.^[6] Owing to their intriguing chemical and physical characteristics, many researchers are looking for new directions to adaptively synthesize 2D or 1D nanomaterials. Among the several classes of nanomaterials,

Group-11 chalcogenide materials have garnered significant research interest in high-performance transistors and sensors due to their exceptionally high carrier mobilities and promising applications in photocatalysis.^[7] Due to advances in experimental fabrication and computational methods, numerous two- and one-dimensional materials composed of group-11 elements, such as copper (Cu), silver (Ag), and gold (Au), combined with different chalcogen elements, have been discovered. Recent studies show that two-dimensional M_2X and MNX (M, N=Cu, Ag, Au; X=S, Se, Te) have been shown to have high stability, auxetic behavior, and excellent electronic properties.^[7h,i,8] The presence of Cu(I), Au(I), and Ag(I) metals in both the M_2X and MNX nanostructures reveals the existence of metalophilic interactions, which play a crucial role in stabilizing the nanosheets.^[8b]

First experimental reported on single-crystal AuSe which exist two distinct crystalline α - and β -phase structures.^[9] Among them the β -AuSe phase has a layered van der Waals (vdW) structure whereas in the α -AuSe phase shows a tube-like structure (Figure 1). Computational studies have revealed that monolayers of β -MX (M=Au, Ag; X=S, Se, Te) exhibit remarkable light-capturing capabilities and hold promise as photocatalysts for oxygen evolution reactions (OER).^[7b] Furthermore, these 2D β -MX monolayers demonstrate in-plane anisotropic behaviour regarding absorption coefficients, photoconductance, and carrier mobility.^[7f,h,10] Similarly, the bulk α -MX phase has been studied in both experiment and computation, which are energetically stable and act as a promising photocatalyst.^[7e,10b,11]

There is a paucity of research on α -MX tube-like nanostructures. In this study, we have used computational techniques to explore the stability, atomic interactions, electronic properties, and optical characteristics of single-strand α -MX nanowires, where M=Au, Ag, and X=S, Se. Additionally, we

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