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Introduction

Excessive consumption of fossil fuels has resulted in accumulation of higher concentrations of CO_2 in the atmosphere, posing a serious challenge to sustainable development on earth.^{1,2} The global estimates of CO_2 are expected to shoot up to 600 ppm by the end of this century, leading to serious environmental and health issues. Thus, mitigating the concentrations of CO_2 and converting it into useful chemicals has become an important research topic. Among the different approaches, electrocatalytic reduction of CO_2 utilizing renewable sources of

Nickel single atom catalyst supported on the gallium nitride monolayer: first principles investigations on the decisive role of support in the electrocatalytic reduction of CO₂[†]

Romana Khanam, Afshana Hassan, Zeeshan Nazir and Manzoor Ahmad Dar 😳 *

Designing efficient and low cost electrocatalysts for the reduction of CO2 to valuable chemicals is a sustainable way of mitigating and balancing its concentrations in the atmosphere which is essential for avoiding effects like climate change and global warming. Herein by means of systematic density functional theory simulations, we investigate the effect of support on the activity/selectivity of Ni based single atom catalyst (SAC) supported on GaN, MoS2, Mo2C, g-C2N and graphyne monolayers towards CO2 activation and reduction to different C1 products. Our results reveal that the Ni SAC strongly binds on all the monolayer supports forming highly stable catalysts. The type of support plays a strong role in tuning the binding and activation of CO2 with Ni SAC supported on GaN and MoS2 monolayers showing the highest CO2 binding energies. Rigorous and in-depth electronic structure analysis reveals that the CO2 binding energy on these catalysts can be successfully rationalized in terms of electronic properties such as the d-band centre and integrated crystal orbital Hamilton populations. Moreover, the computed reaction pathways using the computational hydrogen electrode model indicate that the NI SAC supported on the GaN monolayer can catalyse the CO2 reduction to CH3OH at a record low limiting potential of -0.28 V whereas the Ni SAC supported on the MoS2 monolayer catalyses CO2 reduction to HCOOH at a limiting potential of -0.42 V. Thus, our results show that the nature and type of support plays critical role in modulating the CO2 reduction activity/selectivity on these catalysts and provide insightful guidance for effective catalyst design for CO2 conversion to value added chemicals.

> energy holds great promise to solve the environmental issues related to large CO₂ emissions.³ Towards this direction electrocatalysts based on elemental metals^{4,5} have been tested already for CO₂ reduction.⁶ However, the poor selectivity and low activity due to sluggish and complex reaction kinetics act as bottlenecks for reduction of CO₂ in these systems. To surmount these challenges and to attain improved CO₂ reduction, novel strategies must be developed for the design of modern electrocatalysts.

> Recently, single metal atoms dispersed on suitable twodimensional materials due to their intriguing electronic structure and geometric features have been found to be a promising ground for catalysing several electrocatalytic reactions.⁷⁻¹¹ This has motivated many researchers to investigate single atom catalysts (SACs) particularly based on Cu¹² and Ni¹³ for CO₂ reduction to different value added chemicals.^{14,15}

> Ni-based single atom catalysts have recently garnered considerable interest for the CO₂RR due to their high catalytic efficiency, good selectivity and low overpotentials.^{16,17} To this end, Ni single atoms dispersed over graphene were synthesised by Jiang *et al.*¹⁸ and explored for CO₂RR activity. It was found that these Ni SACs are highly active for CO₂ conversion to CO

Department of Chemistry, Islamic University of Science and Technology, Awantipora, Jammu and Kashmir 192122, India. E-mail: manzoor.dar@islamicuniversity.edu.in † Electronic supplementary information (ESI) available: Calculated dissolution potentials, spin polarized orbital projected density of states (oPDOS) of Ni SACs supported on different monolayers, top view of CO₂ adsorbed on Ni SACs supported on g-C₂N, GaN, graphyne, Mo₂C and MoS₂, charge density difference plot of CO₂ adsorbed on Ni SACs supported on GaN, MoS₂, and Mo₂C monolayers, d-band center results and optimized structures of CO₂RR intermediates on Ni SACs supported on GaN, MoS₂ and Mo₂C, AIMD simulation of Ni SACs supported on GaN and energy, ZPE, TS and G of reaction steps of the CO₂RR on nickel SACs supported on GaN, MoS₂ and Mo₂C catalysts. See DOI: https://doi.org/10.1039/d3se00830d