



Structural analysis and theoretical studies of 2-(2-chlorobenzylidene) malononitrile (CS) “a riot controlling agent”

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ABSTRACT

In solvent methanol, malononitrile and 2-chlorobenzaldehyde were stirred in the presence of base piperidine to synthesise the target chemical 2-(2-chlorobenzylidene) malononitrile (CS). The crystal structure of the target compound $C_{10}H_5ClN_2$ has been elucidated through X-ray crystallographic analysis. It crystallises in the monoclinic crystal system within the P21 space group, containing four molecules per asymmetric unit. The unit cell dimensions are $a = 3.8825(3) \text{ \AA}$, $b = 21.0202(18) \text{ \AA}$, $c = 21.3481(18) \text{ \AA}$, and $\beta = 95.096(4)^\circ$. The phenyl and malononitrile groups, which make up the molecular components, are each planar but slanted towards one another at a dihedral angle of $12.7^\circ (4)$. Besides this, target molecule is also stabilized by hydrogen bonding, halogen bonding and π stacking interactions. Solid-state structure has been analysed through Hirshfeld surface analysis, including the evaluation of the different energy frameworks, indicating that the molecular sheets are primarily formed by hydrogen bonds and halogen bonds and the stabilization is dominated via the electrostatic energy contribution. Fingerprint plots were carried out to study the relative contribution of noncovalent interactions in the target compound CS. Further, the density functional theory calculations were employed using B3LYP hybrid functional with 6-311++G level basic set to optimize the structural coordinates in gas phase and in different solvents (chloroform, water, and methanol). The chemically active regions of the target compound were identified by the analysis of molecular electrostatic potential surface. Further, from PASS analysis and literature reports, molecular docking of title compounds was carried out with human transient receptor potential ankyrin 1 ion channel (hTRPA1) (PDB-ID 6PQO). ΔG_b (docking energy) and inhibition constant was calculated to be -7.53 kcal/mol and 3.33 \mu M respectively, with the target ion channel.

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